

# Desalination of seawater by direct sunlight

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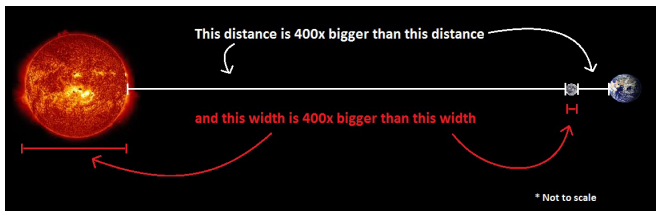
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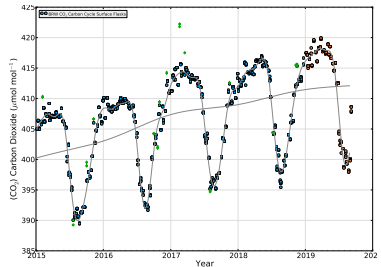
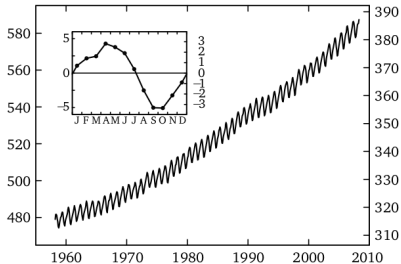
# Problem statement

- Intro + Objectives
  - Global warming: What and Why
- Black-body radiation (better to think “microwave”)
- Sun ( $T=7000^{\circ}\text{C}$ ) vs.  $1.3\text{ [kW/m}^2\text{]}$  on Earth ( $T=25^{\circ}\text{C}$ )



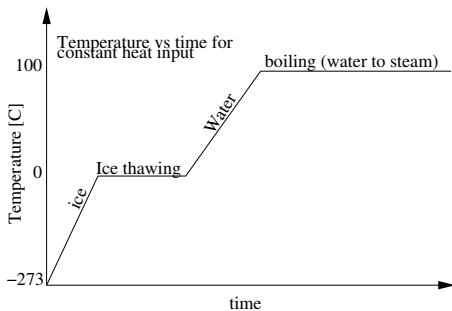
- Earth: More than @400 [ppm] of  $\text{CO}_2$  tips the energy balance
- **Mars:**  $\text{CO}_2$  (95%) and an ocean of  $\text{H}_2\text{O}$  a *billion* years ago  
Water on Mars today? Gone! **[Chang 2021]**

# CO<sub>2</sub> levels since 1950 (60 and 5 years)



- 1 What is the source of climate change?
  - The CO<sub>2</sub> increases by 1.667 ppm each year?
  - It has increased 100 ppm in 60 years
  - The slope has increased to 2 ppm in the last 20 years
  - The CO<sub>2</sub> oscillate, decreasing in summer, **increasing in winter.**
- 2 CO<sub>2</sub> is the main source of “climate change”
- 3 CO<sub>2</sub>(H<sub>2</sub>CO<sub>3</sub>) accounts for the (↑) ocean acidity (PH ↓)

## States of water with a constant solar heat input $Q_o$



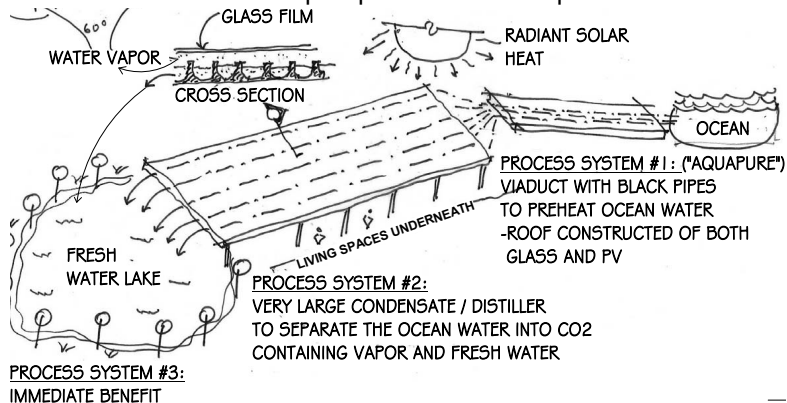
- For ice:  $T(t) = c_p Q_o (T \text{ [}^\circ\text{C]}, Q_o \text{ [heat energy input/sec]})$
- For thawing:  $\Delta Q = L_o Q_o$  and  $T = 0 \text{ [}^\circ\text{C]}$  (is constant)
- Once all the ice is thawed,  $T \uparrow$  & the water begins to evaporate
- When boiling,  $T = 100 \text{ [}^\circ\text{C]}$  (is constant)

# How to proceed?

- “ We don't have enough water?” We have the oceans!
- Massive desalination is needed
  - 1 Use solar warming to evaporate water ( $T \uparrow$ ) and seawater's cold to condense the vapor ( $T \downarrow$ )
  - 2 This  $\uparrow\downarrow$  cycle can produce huge amounts of water at low cost
  - 3 Goal: Sun as the energy source  $\Rightarrow$  seawater  $\rightarrow$  pure water
  - 4 But HOW???

# Sunlight (solar radiant heat) to pure H<sub>2</sub>O, direct

-The cold ocean water is pumped into the evaporation chamber.

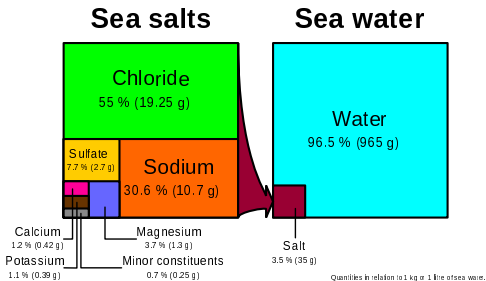


-The 60 [°C] vapor is *condensed* to 20 [°C], warming the seawater

-Explain how this works (Allen 2021) [Video]

Questions? (<https://www.youtube.com/watch?v=009qvgRoiLo>)

# Composition of seawater



- The  $\text{CO}_2$  is only 0.04%, which determines the PH, and the destruction of coral reefs (life in the ocean).
- Evaporation gives pure water, leaving the salt behind
- The water can be used to grow plants in the desert, which consumes the  $\text{CO}_2$ .
- You can't grow plants in salty seawater!!!

# Massive desalination: How?

- 1 CO<sub>2</sub>@415 [ppm] tips earth's delicate energy balance
- 2 We can use direct sunlight to heat & evaporate cold seawater.
- 3 How?
  - The sun heats the seawater to 5% vapor (dew-point 40 °C)
  - Water vapor is pure (0% salt)
  - The water vapor is then cooled using the cold (15 °C) seawater
  - Cooling the water vapor releases the vapor latent heat.

This raise the water temperture further, increasing the yield!

Condensation using cold seawater ⇒ maximum yield!

- 4 Seawater  $\xrightarrow{\text{sun}}$  vapor (93%) + salt (3.5%) + H<sub>2</sub>O (7%)  
1 kgm  $\xrightarrow{\text{sun}}$  H<sub>2</sub>O
- Seawater → pure water + brine (100% solar powered)
- No inefficient solar panels (>77% ≈ 4/5 energy loss).
- No power-hungry RO, or clogged membranes
- ultra **GREEN**



# The key question is the recovery of the latent heat

- The main question is: *How much of the heat of vaporization from the condensed vapor, may be recovered?*
  - First the sun heats the water to the dew-point  $\Rightarrow$  5% vapor
  - Vapor is cooled using the cold seawater (via a heat-exchanger), heating it even more than the daytime solar input.
  - Thus less sunlight is needed to reach  $T_{\max}$  (40-60 [°C])
  - 
  - Given a proper design, and assuming (e.g. 15 °C) inlet seawater (and -3.9 [°C] night air), the  $T \uparrow, \downarrow$  efficiency could be >95%
  - The air temperature in the Sahara desert at night is below freezing, so the water vapor could be iced:  
*“During the day, desert temperatures rise to an average of 38°C (a little over 100°F). At night, Sahara temperatures fall to an average of -3.9°C (about 25°F)”*

Concept: Flow the vapor under the inlet seawater, exposing it to the inlet seawater (or night air), cooling (or iceing) the vapor, further heating the seawater.

# History of the Railroad & Auto industries

- Who will pay for it?
- The billionaires (c1870) in the United States made their money on the railroads
- There was a huge need for inexpensive transportation
- Likewise, today there is a huge need for cheap clean water
- The water industry today and the transportation industry have a great deal in common.
- There is huge opportunity here: Clean water  $\equiv$  “mega-bucks.”
- A business model is needed to proceed.

Questions? Comments? Thoughts?

# Bibliography

Wang et al. 2019

Chang 2021

Allen 2020